

## Physics

1 (a)

$$\text{Let } F \propto P^x V^y T^z$$

By substituting the following dimensions:

$$[P] = [ML^{-1}T^{-2}], [V] = [LT^{-1}], [T] = [T]$$

and comparing the dimension of both sides

$$x = 1, y = 2, z = 2, \text{ so } F = PV^2T^2$$

2 (a)

$$\text{Volume of cube} = a^3$$

$$\text{Surface area of cube} = 6a^2$$

$$\text{according to problem } a^3 = 6a^2 \Rightarrow a = 6$$

$$\therefore V = a^3 = 216 \text{ units}$$

4 (b)

5 (b)

6 (c)

$$CR \text{ is known as time constant } CR = [T]$$

7 (a)

Quantities having different dimensions can only be divided or multiplied but they cannot be added or subtracted

8 (a)

$$B \text{ is unitless. } \therefore \text{Unit of } B \text{ is } m^{-1}s^{-1}$$

10 (b)

$$mv = kg \left( \frac{m}{\text{sec}} \right)$$

11 (c)

12 (b)

Time constant in an  $R - C$  circuit

$$\tau = R - C$$

$$[\tau] = [R][C]$$

$$= [ML^2T^{-3}A^{-2}][M^{-1}L^{-2}T^4A^2]$$

$$= [M^0L^0T]$$

13 (d)

$$R = 8.3 \text{ J/K-mol}$$

$$n_1 u_1 = n_2 u_2$$

$$\therefore n_2 = \frac{n_1 u_1}{u_2}$$

$$= \frac{8.3 \text{ J/K-mol}}{\text{atm L/K-mol}}$$

$$= \frac{8.3 \text{ J/K-mol}}{8.3 \text{ J/K-mol}}$$

$$= \frac{8.12}{(1.013 \times 10^5 \text{ N/m}^2)(10^{-3} \text{ m}^3)/\text{K-mol}}$$

$$= \frac{8.12}{10^2} = 0.0812$$

$$\therefore 8.3 \text{ J/K-mol} = 0.0812 \text{ atm L/K-mol}$$

14 (a)

From Coulomb's law

$$F = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2}$$

$$\begin{aligned} \Rightarrow \left[ \frac{1}{4\pi\epsilon_0} \right] &= \frac{[F \times r^2]}{[q]^2} \\ &= \frac{[\text{newton}][\text{metre}]^2}{[\text{coulomb}]^2} \\ &= \text{Nm}^2\text{C}^{-2} \end{aligned}$$

15 (a)

Required percentage error

$$= 2 \times \frac{0.01}{15.12} \times 100 + \frac{0.001}{10.15} \times 100 = 4 + 1 = 5$$

## Chemistry solutions

1 (b)  
Ar and  $\text{Ca}^{2+}$  are isoelectronic species as they have same number of electrons, *i. e.*, 18.

2 (a)  
Tritium is the isotope of hydrogen. Its composition is as follows :  
1 electron, 1 proton and 2 neutrons

3 (b)  
 $E = \frac{hc}{\lambda}$ ,  $h$  and  $c$  for both causes are same so,  
$$\frac{E_1}{E_2} = \frac{\lambda_2}{\lambda_1} = \frac{16000}{8000}$$
$$E_1 = 2E_2$$

4 (c)  
When  $n = 4$  and  $x = 5$  then electronic configuration can be written as  
 $(4 - 1)s^2(4 - 1)p^6(4 - 1)d^54s^2$   
This electronic configuration represents Mn and its atomic number is 25. Hence,  
number of protons are 25 in its nucleus.

5 (c)  
Isotones are species which have equal number of neutrons.  
Neutrons in  ${}_{19}\text{K}^{39} = 39 - 19 = 20$   
Neutrons in  ${}_{20}\text{Ca}^{40} = 40 - 20 = 20$

6 (d)  
The mass number = atomic number + number of neutron  
Atomic number = no. of proton  
= no. of electron (for an atom)  
So, mass number =  $18 + 20 = 38$

7 (c)  
Bohr's theory is applicable to unielectron atom or ion only.

8 (a)  
Wave number of spectral line in emission spectrum of hydrogen,

$$\bar{\nu} = R_H \left( \frac{1}{n_1^2} - \frac{1}{n_2^2} \right) \quad \dots (i)$$

$$\text{Given, } \bar{\nu} = \frac{8}{9} R_H$$

On putting the value of  $\bar{\nu}$  in Eq. (i), we get

$$\frac{8}{9} = R_H \left( \frac{1}{n_1^2} - \frac{1}{n_2^2} \right)$$

$$\frac{8}{9} = \frac{1}{(1)^2} - \frac{1}{n_2^2}$$

$$\frac{8}{9} - 1 = -\frac{1}{n_2^2}$$

$$\frac{1}{3} = \frac{1}{n_2}$$

$$\therefore n_2 = 3$$

Hence, electron jumps from  $n_2 = 3$  to  $n_1 = 1$

9 (d)  
Angular momentum of an electron

$$= mvr = \frac{nh}{2\pi} \text{ (n is orbit number)}$$

$$\text{in 5th orbit} = \frac{5h}{2\pi} = \frac{2.5h}{\pi}$$

10 (d)

Hydrogen spectrum is an emission spectrum. It shows the presence of quantized energy levels in hydrogen atom.

11 (d)

$$\text{Bohr radius for } n\text{th orbit} = 0.53 \times \frac{n^2}{Z}$$

Where,  $Z$  = atomic number

$$\therefore \text{Bohr radius of 2nd orbit of } \text{Be}^{3+} = \frac{0.53 \times (2)^2}{4}$$

$$= 0.53 \text{ \AA}$$

$$\text{(d) Bohr radius of 1st orbit of H} = \frac{0.53 \times (1)^2}{1}$$

Hence, Bohr's radius of 2nd orbit of  $\text{Be}^{3+}$  is equal to that of first orbit of hydrogen.

12 (a)

$\lambda$  for visible light is in the range of 400 to 780 nm.

$$E = \frac{hc}{\lambda}$$

This, it is in the range of electron volt (eV).

13 (a)

The value of Rydberg constant is  $10,9678 \text{ cm}^{-1}$ .

14 (d)

According to Bohr model,

Radius of hydrogen atom

$$(r_n) = \frac{0.529 \times n^2}{Z} \text{ \AA}$$

Where,  $n$  = number of orbit

$Z$  = atomic number

$$r_3 = \frac{0.529 \times (3)^2}{1} = 4.761 \text{ \AA}$$

15 (c)

$$\text{Kinetic energy} = \frac{1}{2}mv^2,$$

$$\text{Potential energy} = \frac{-e^2}{r}$$

$$\text{But, } mv^2 = \frac{e^2}{r}$$

$$\text{KE} = \frac{1}{2} \frac{e^2}{r}$$

Total energy = KE + PE

$$= \frac{1}{2} \frac{e^2}{r} - \frac{e^2}{r} = \frac{e^2}{r} \left( \frac{1}{2} - 1 \right) = \frac{-e^2}{2r}$$

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Hints and solution 11<sup>th</sup> Science Logarithm

Q.31  $\log_{2\sqrt{2}} 32\sqrt{4} = \log_{(2^{3/2})} (2^5 \cdot 4^{1/2}) = \log_{2^{3/2}} (2^{5+\frac{2}{2}}) = \frac{2}{3} \cdot \frac{27}{5} \log_2 2$   
 $= \frac{2}{3} \cdot \frac{27}{5} = \frac{18}{5} = 3.6$  Ans option (a)

Q.32. given  $x = \log_3 5$  |  $y = \log_{17} 25 = \log_{17} 5^2$   
 $\Rightarrow \frac{1}{x} = \log_5 3$  |  $\therefore \frac{1}{y} = \frac{1}{2} \log_5 17$   
 $\Rightarrow \frac{1}{x} = \frac{1}{2} \log_5 9$

$\therefore \frac{1}{y} > \frac{1}{x} \Rightarrow x > y$  Ans option (b)

Q.33  $y = 2^{\frac{1}{\log_4 x}}$

$\Rightarrow \log_2 y = \frac{1}{\log_4 x} = \log_4 x = \frac{1}{2} \log_2 x$

$\therefore 2 \log_2 y = \log_2 x$

$\Rightarrow \boxed{y^2 = x}$  Ans option (a)

Q.34.  $\log_2 (3x-2) = \log_{1/2} x$

$\Rightarrow \log_2 (3x-2) = -\log_2 x = \log_2 x^{-1}$

$\Rightarrow (3x-2) = x^{-1}$

$\Rightarrow 3x^2 - 2x = 1 \Rightarrow x = 1$  or  $\frac{1}{3}$   
 $\checkmark$  Ans

But given question are meaningful if  $x > \frac{2}{3}$ .

Hence Ans option (d)

Q.35  $\log_3(x-2) \leq 2$   
 $[\because (x-2) > 0]$

$\Rightarrow 0 < (x-2) \leq 3^2$

$\Rightarrow 2 < x \leq 9+2$

$\Rightarrow \boxed{2 < x \leq 11}$

option (b)

Q.36  $\log_{0.2}|x-3| \geq 0$

$\Rightarrow 0 < |x-3| \leq (0.2)^0$

$\Rightarrow -1 \leq x-3 \leq 1$  &  $x-3 \neq 0$

$\Rightarrow \boxed{2 \leq x \leq 4}$  &  $x \neq 3$

$\therefore x \in [2, 4] - \{3\}$

option (b)

Q.37  $6(\log_x 2 - \log_4 x) + 7 = 0$

$\Rightarrow 6(\log_2 2 - \frac{1}{2} \log_2 x) + 7 = 0$

$\Rightarrow 6(\frac{1}{2} - \frac{1}{2} \cdot y) + 7 = 0$

$\Rightarrow 6(\frac{2-y^2}{2y}) + 7 = 0$

$\Rightarrow 6 - 3y^2 + 7y = 0$

$\Rightarrow 3y^2 + 2y - 9y - 6 = 0$

$\Rightarrow (y-3)(3y+4) = 0$

$\Rightarrow y = 3$  or  $-2/3$

put,  $\log_2 x = y$

$\therefore \log_x 2 = \frac{1}{y}$

$\therefore \log_2 x = 3$  or  $\log_2 x = -2/3$

$\therefore x = 2^3$

$\boxed{x = 8}$

or  $\boxed{x = 2^{-2/3}}$

option (c)

Q.38.  $\log_2(x-1) > 4$

$$\Rightarrow (x-1) > 2^4$$

$$\Rightarrow \textcircled{x > 17} \text{ Ans} \quad \underline{\text{option (c)}}$$

Q.39  $\log_4(2 \times 4^{x-2} - 1) + 4 = 2x$

$$\Rightarrow \log_4(2 \times 4^{x-2} - 1) = 2x - 4$$

$$\Rightarrow 2 \times 4^{x-2} - 1 = 4^{2x-4} = 4^{2(x-2)} = (4^{x-2})^2$$

$$\Rightarrow 2y - 1 = y^2$$

$$\Rightarrow y^2 - 2y + 1 = 0$$

$$\Rightarrow (y-1)^2 = 0 \Rightarrow \textcircled{y=1}$$

$$\therefore 4^{x-2} = 1 = 4^0$$

$$\Rightarrow x-2 = 0$$

$$\Rightarrow \textcircled{x=2} \text{ Ans} \quad \underline{\text{option (c)}}$$

Q.40  $\log_{0.5} \frac{3-x}{x+2} < 0$

$$\Rightarrow \frac{3-x}{x+2} > (0.5)^0$$

$$\Rightarrow \frac{3-x}{x+2} > 1$$

$$\Rightarrow \frac{3-x-x-2}{x+2} > 0$$

$$\Rightarrow \frac{-2x+1}{x+2} > 0$$

$$\Rightarrow \frac{2x-1}{x+2} < 0$$

$$\therefore 2x-1 < 0 \text{ and } x+2 > 0$$

$$\Rightarrow \textcircled{x < \frac{1}{2}} \text{ and } \textcircled{x > -2}$$

$$\therefore \boxed{-2 < x < \frac{1}{2}} \text{ Ans}$$

$$\underline{\text{option (b)}}$$

Single Correct Answer Type

41 (2)

Since,  $A + B + C = \pi$

$$\Rightarrow a = \pi - (B + C)$$

We have,  $\cos A = \cos B \cos C$

$$\Rightarrow \cos[\pi - (B + C)] = \cos B \cos C$$

$$\Rightarrow -\cos(B + C) = \cos B \cos C$$

$$\Rightarrow -[\cos B \cos C - \sin B \sin C] = \cos B \cos C$$

$$\Rightarrow \sin B \sin C = 2 \cos B \cos C$$

$$\Rightarrow \tan B \tan C = 2$$

42 (1)

$$x^2 + y^2 + z^2 = r^2 \sin^2 \theta \cos^2 \phi + r^2 \sin^2 \theta \sin^2 \phi + r^2 \cos^2 \theta$$

$$= r^2 \sin^2 \theta (\cos^2 \phi + \sin^2 \phi) + r^2 \cos^2 \theta$$

$$= r^2 \sin^2 \theta + r^2 \cos^2 \theta$$

$$= r^2$$

43 (3)

We have,  $x \cos \theta = y \cos \left( \theta + \frac{2\pi}{3} \right)$

$$= z \cos \left( \theta + \frac{4\pi}{3} \right) = k \quad (\text{say})$$

$$\Rightarrow \cos \theta = \frac{k}{x}, \cos \left( \theta + \frac{2\pi}{3} \right) = \frac{k}{y}$$

$$\text{and } \cos \left( \theta + \frac{4\pi}{3} \right) = \frac{k}{z}$$

$$\therefore \frac{k}{x} + \frac{k}{y} + \frac{k}{z} = \cos \theta + \cos \left( \theta + \frac{2\pi}{3} \right) + \cos \left( \theta + \frac{4\pi}{3} \right)$$

$$= \cos \theta - \cos \left( \frac{\pi}{3} - \theta \right) - \cos \left( \frac{\pi}{3} + \theta \right)$$

$$= \cos \theta - 2 \cos \frac{\pi}{3} \cos \theta = 0$$

$$\Rightarrow \frac{1}{x} + \frac{1}{y} + \frac{1}{z} = 0$$

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(3)

Given that,  $\sin \theta + \cos \theta = m \dots$  (i)and  $\sec \theta + \operatorname{cosec} \theta = n \dots$  (ii)Now,  $n(m+1)(m-1) = n(m^2-1)$  $= (\sec \theta + \operatorname{cosec} \theta) 2 \sin \theta \cos \theta \quad (\because m^2 = 1 + 2 \sin \theta \cos \theta)$ 

$$= \frac{\sin \theta + \cos \theta}{\sin \theta \cos \theta} 2 \sin \theta \cos \theta$$

$$= 2m \quad [\text{from Eq.(i)}]$$

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(1)

We have,  $\tan(\alpha + \beta) = \frac{\tan \alpha + \tan \beta}{1 - \tan \alpha \tan \beta}$ 

$$= \frac{\frac{1}{\sqrt{x(x^2+x+1)}} + \frac{\sqrt{x}}{\sqrt{x^2+x+1}}}{1 - \frac{1}{\sqrt{x(x^2+x+1)}} \cdot \frac{\sqrt{x}}{\sqrt{x^2+x+1}}}$$

$$= \frac{(1+x)\sqrt{x^2+x+1}}{\sqrt{x} \cdot x(x+1)}$$

$$= \sqrt{x^{-3} + x^{-2} + x^{-1}} = \tan \gamma \quad (\text{given})$$

$$\therefore \alpha + \beta = \gamma$$